

# Progress in Excess Power Production by Laser Triggering

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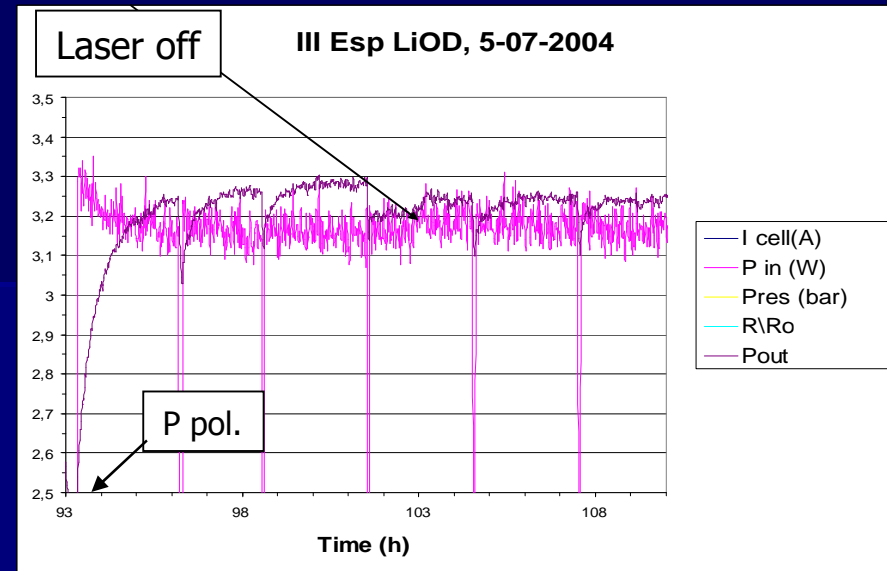
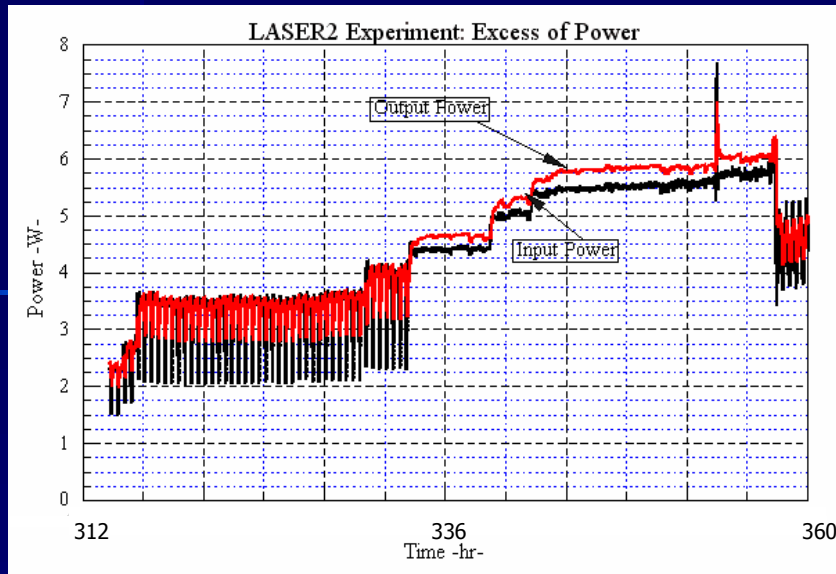
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*(5) Energetics (USA-Israel)*

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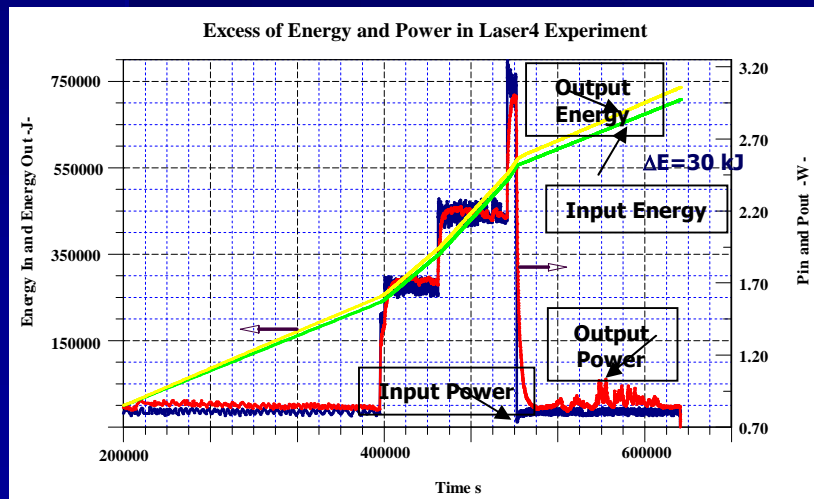
- Previous results
- New cell concept for isoperibolic calorimetry with trigger
- Modelling & engineering
- Calibration
- Results
- Conclusions

## Previous Results

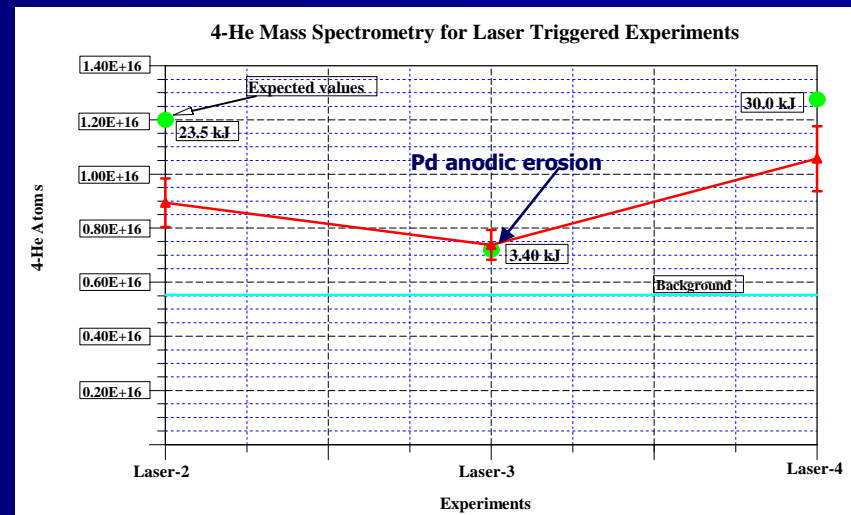


Evolution of the input and output power, last 300 hr under laser irradiation (P-polarization)

Laser on-off effect



Laser 4 experiment

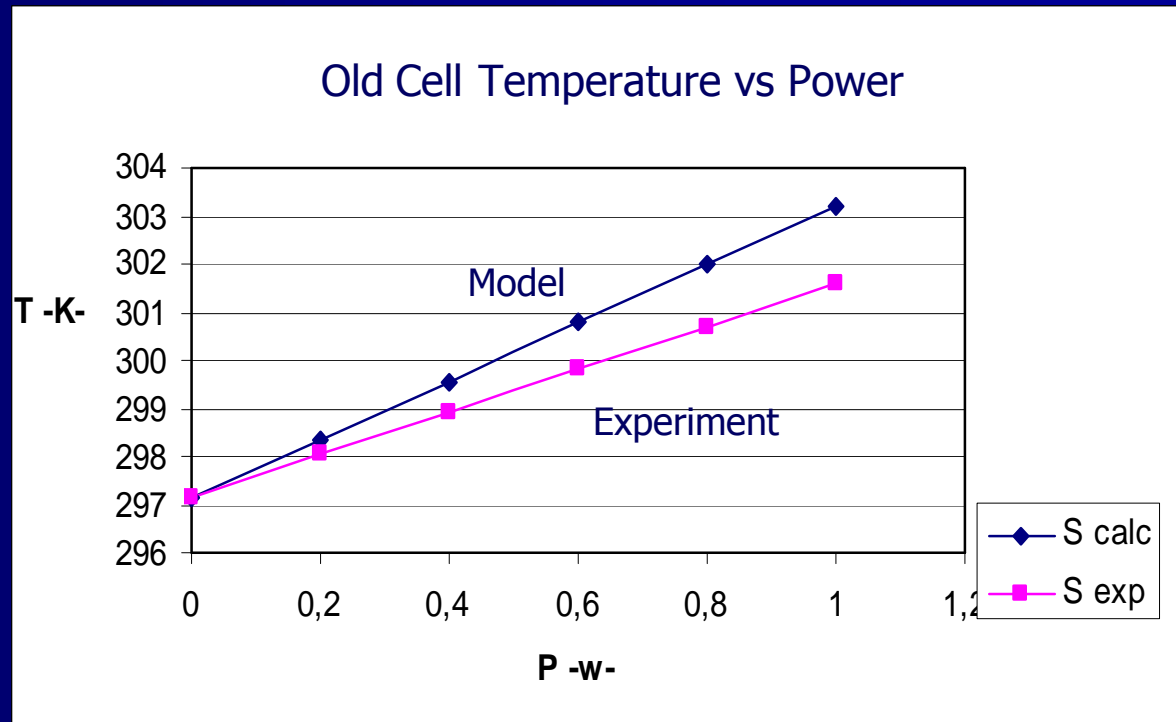


$^4\text{He}$  Production

# Improvement of the Cell

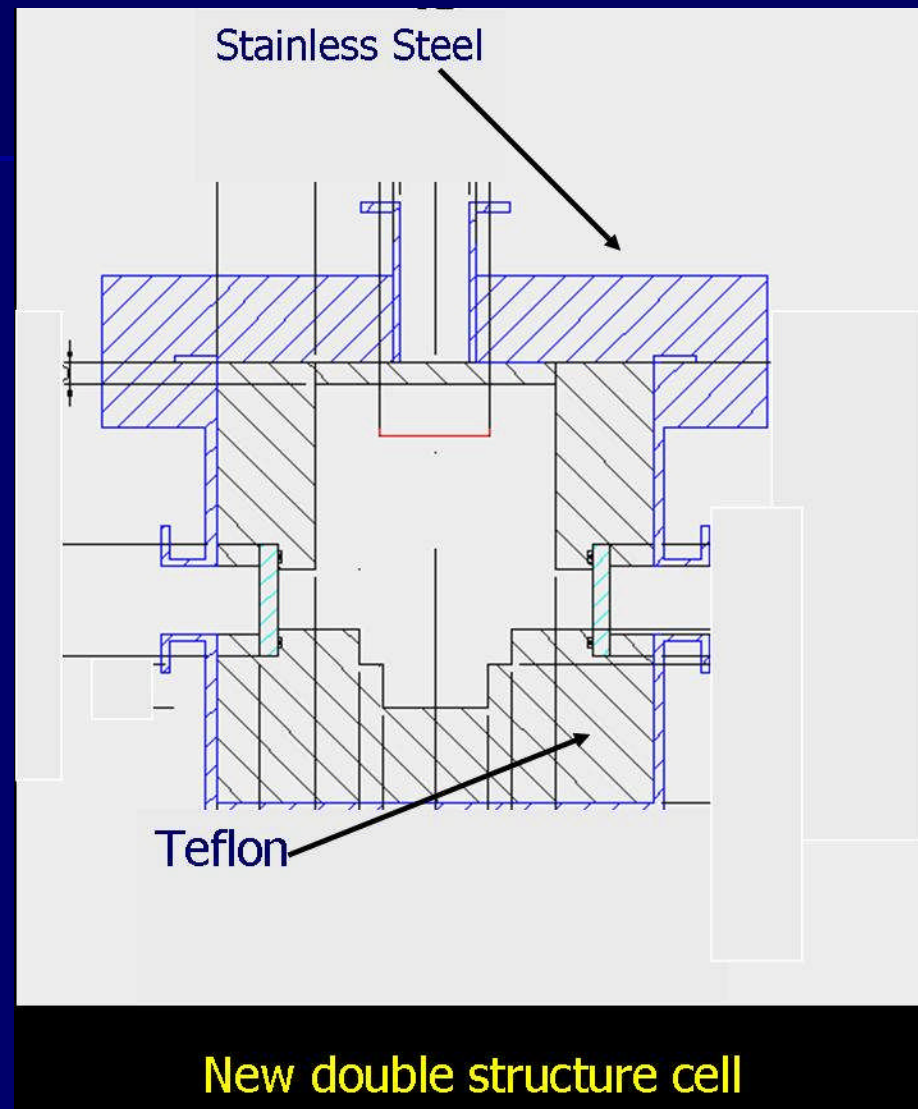


Old cell



Average temperature of the electrolyte vs power

## Double Structure New Cell for Laser Experiments



# Thermal Analysis

## Heat transfer equation

$$\text{div} (K \text{ grad}(T)) + Q = \rho c_p \frac{\partial T}{\partial t} + \rho c_p (V_x \frac{\partial T}{\partial x} + V_y \frac{\partial T}{\partial y} + V_z \frac{\partial T}{\partial z})$$

## Boundary conditions

$$-K \frac{\partial T}{\partial n} = h(T - T_a)$$

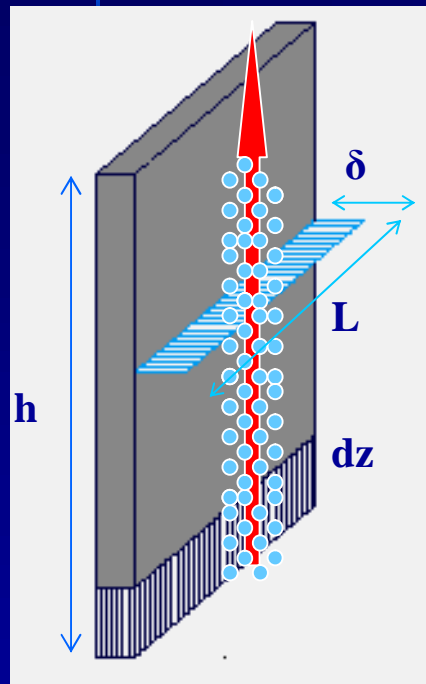
*(convective heat exchange mechanism)*

## Assumptions:

- 3D transient
- isotropic ( $K_x=K_y=K_z$ )
- Steady state boundary conditions (thermostatic box)  $T_{\text{amb}} = \text{const.}(t)$
- Radiative heat exchange negligible

# Hydrogen Bubbles at the Cathode

Gas velocity  $V_{H_2}$  is calculated by means of the current density



Pd foil

Gas flow rate for unit area

$$K = \frac{J \cdot 22,4 \cdot 1000}{nF} \quad (cm/s)$$

Total gas flow rate

$$W = \int_0^z L \cdot K \cdot dz = L \cdot K \cdot z \quad (cm^3/s)$$

Gas velocity

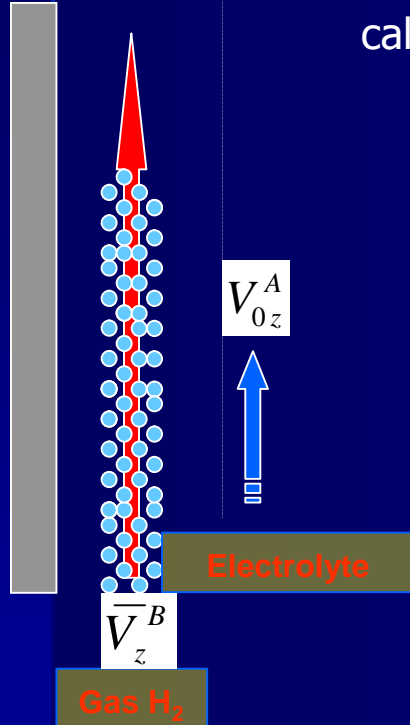
$$V_{gas}(z) = \frac{W}{A} = \frac{W}{L \cdot \delta} = \frac{K \cdot z}{\delta} \quad (cm/s)$$

# Liquid-Gas Interface Velocity

One fluid is moved by the other having different density and viscosity

Liquid-Gas Interface

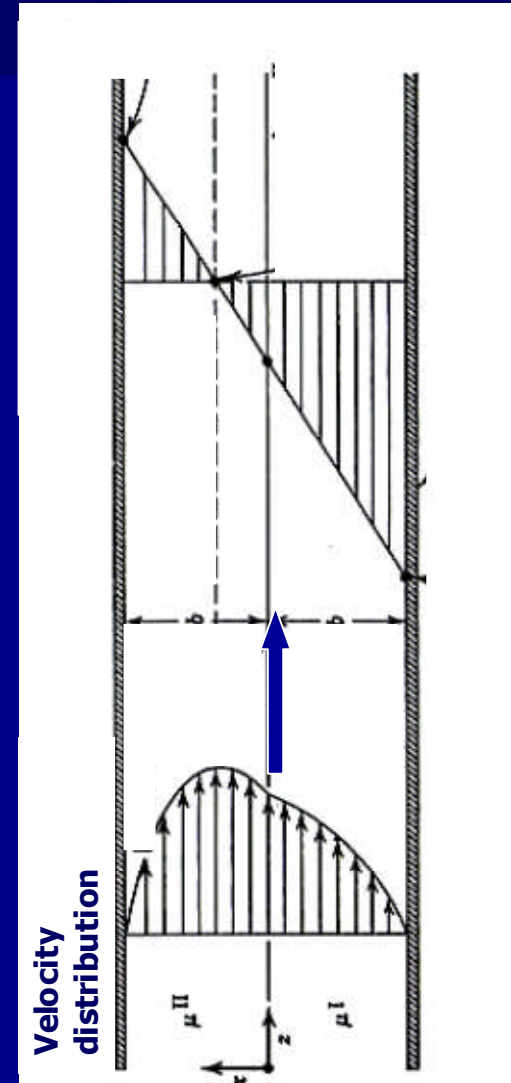
The electrolyte interface velocity is calculated by the average gas velocity



$$V_{H_2Oz}^A = 0,093 \cdot \bar{V}_{H_2z}^B$$


$$\left. \frac{V_z^A}{V_z^B} \right|_{x=0} = \frac{\frac{1}{\mu^A} \left( \frac{2\mu^A}{\mu^A + \mu^B} \right)}{\frac{1}{\mu^B} \left( \frac{2\mu^B}{\mu^A + \mu^B} \right)} = 1$$

$$\frac{V_{0z}^A}{\bar{V}_z^B} = \frac{12\mu^B}{\mu^A + 7\mu^B} = 0.093$$





## Equations


$$\frac{1}{r} \cdot \frac{\partial}{\partial r} \left( r \frac{\partial v_z}{\partial r} \right) = 0$$

**Motion Equation**

$$\frac{1}{r} v_r + \frac{\partial v_r}{\partial r} + \frac{\partial v_z}{\partial z} = 0$$

**Continuity equation**

## Assumptions

axial symmetry

density and viscosity are constant

steady state

negligible effect of pressure and mass

## Iterative procedure for velocity field calculation

### step1: Equation of motion

solving the equation of motion with the boundary conditions

$$v_z = V_{0z} \quad r=a ; \quad v_z=0 \quad r=R$$



$$v_z = -\ln R \cdot \frac{V_o}{\ln\left(\frac{a}{R}\right)} + \frac{V_o \cdot \ln r}{\ln\left(\frac{a}{R}\right)}$$

### Step 2: Continuity equation

by replacing the  $v_z$  calculated at step 1 and by solving the continuity equation **with the condition of zero flow rate** in the radial direction



$$v_r(r) = \frac{C_{1R}}{r} - \frac{K \cdot r \cdot [2\ln\left(\frac{r}{R}\right) - 1]}{4 \cdot \delta \cdot \ln\left(\frac{a}{R}\right)}$$

$$C_{1R} = -\frac{1}{4} \frac{K}{\delta} \cdot \frac{a^2 - R^2 + a^2 \ln\left(\frac{R}{a}\right)}{[\ln(R/a)]^2}$$

### Step 3: Continuity equation

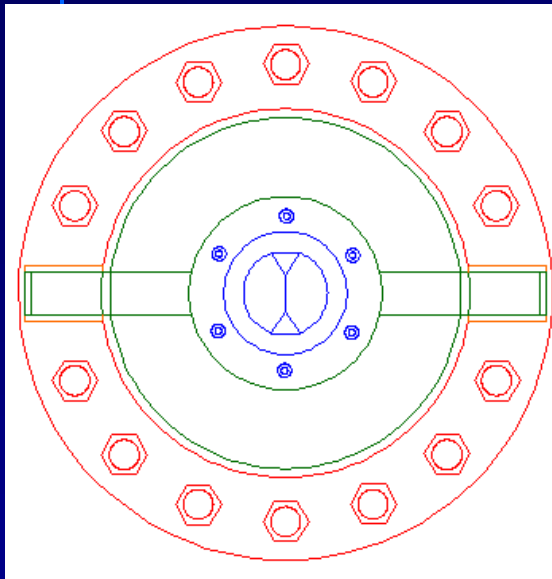
By replacing  $v_r$  calculated at step 2 and by solving the continuity equation with the condition of zero flow rate in axial direction



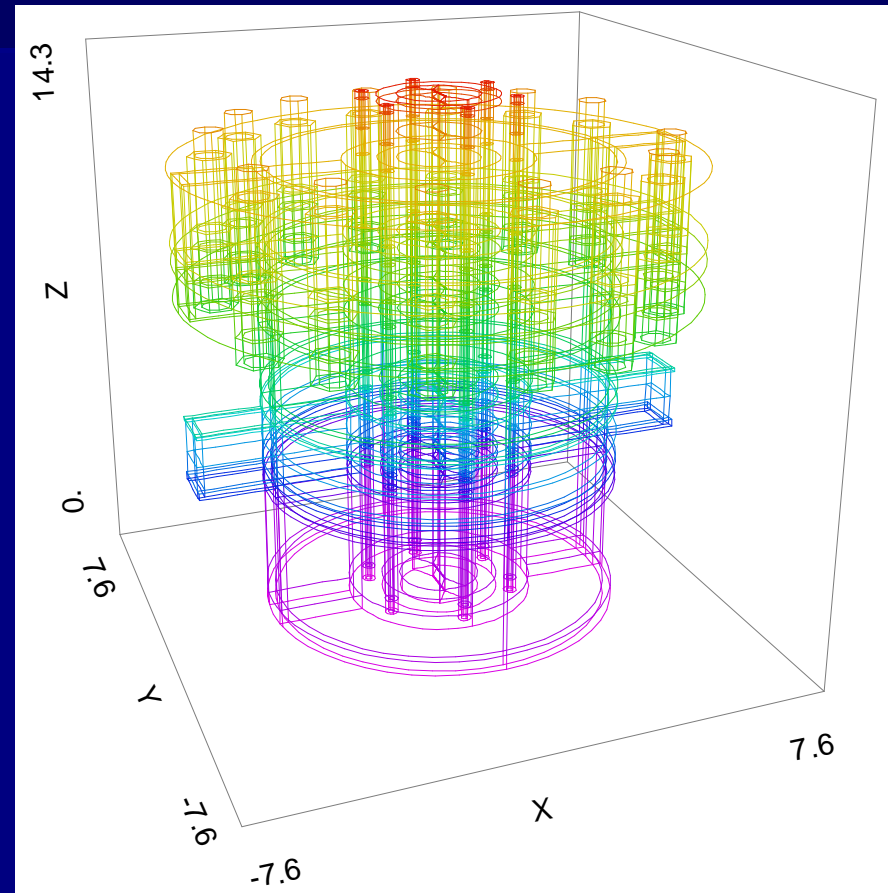
$$VV_z(r, z) = -z \cdot K_2 \cdot \ln\left(\frac{r}{R}\right) + C_{1z}$$

$$C_{1z} = \frac{1}{2} K_2 \cdot z \cdot \frac{(a^2 - R^2 + 2\ln R/a)}{a^2 - R}$$

# FEM Domain

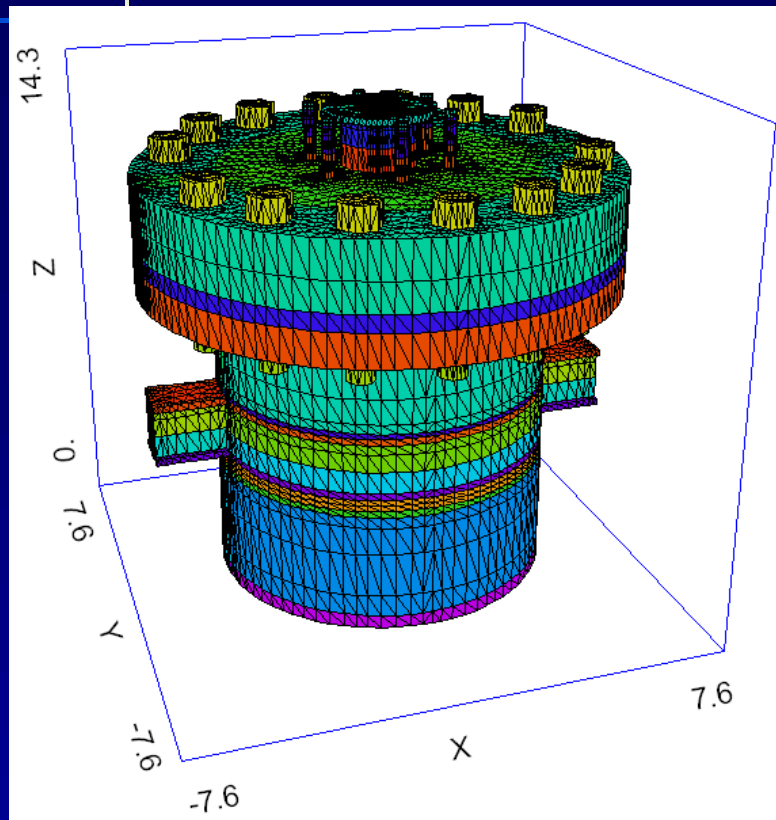


CAD 2D

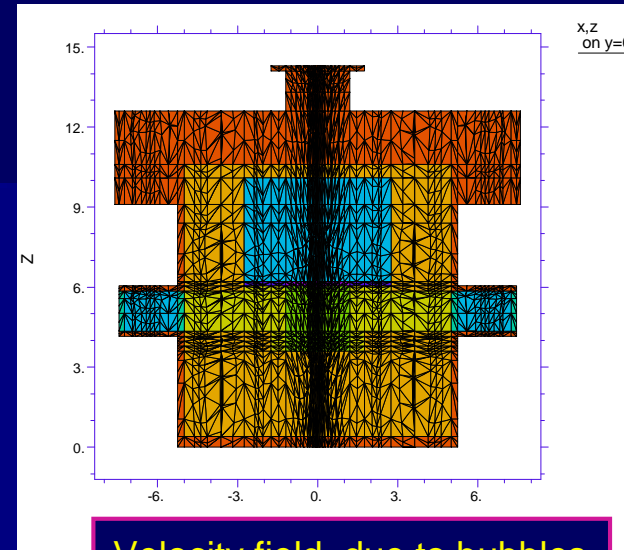
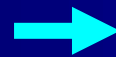


Domain estrusion

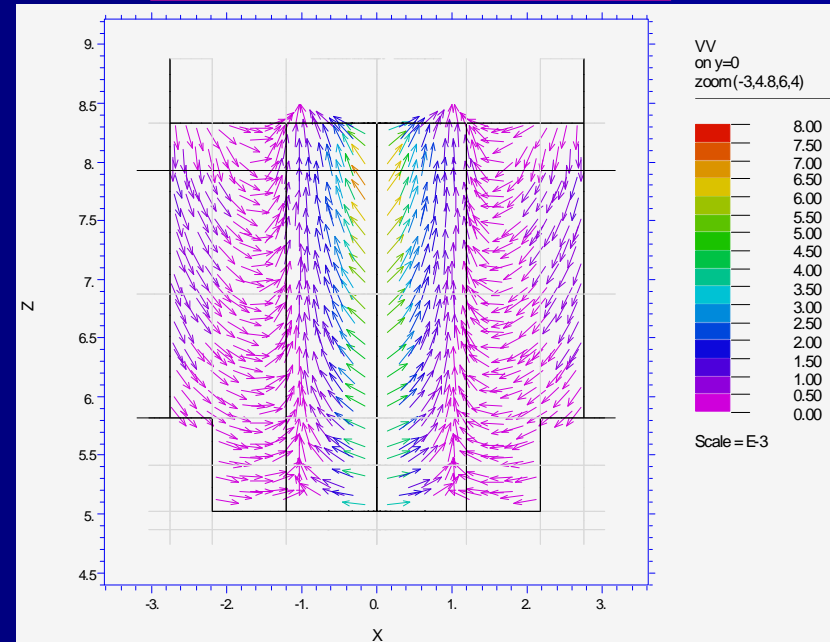
# Mesh and Velocity Field



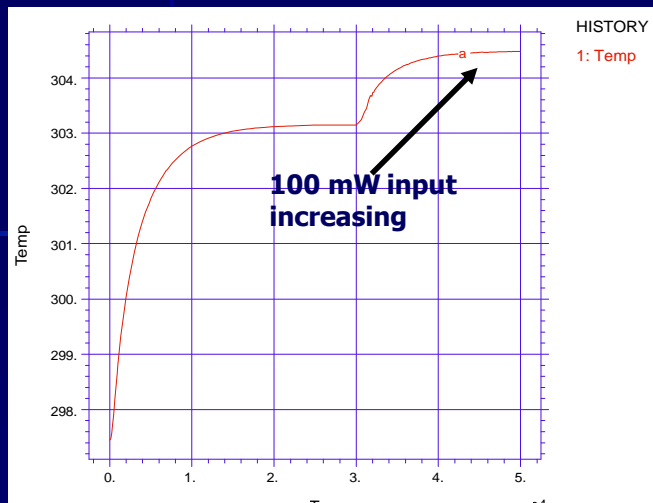
Mesh 3D



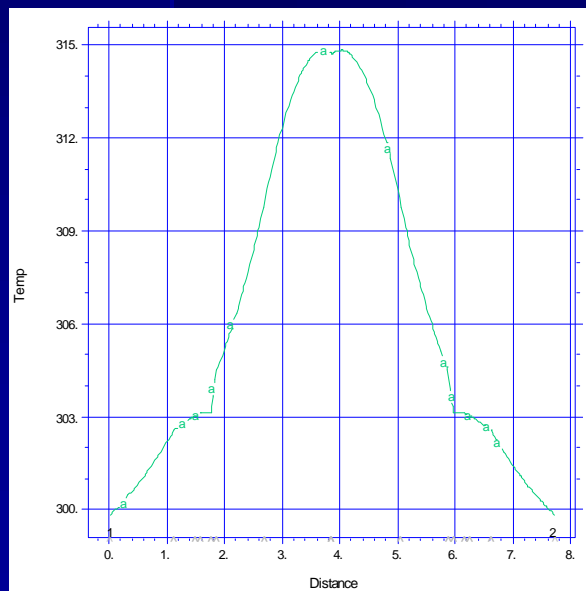
Velocity field due to bubbles



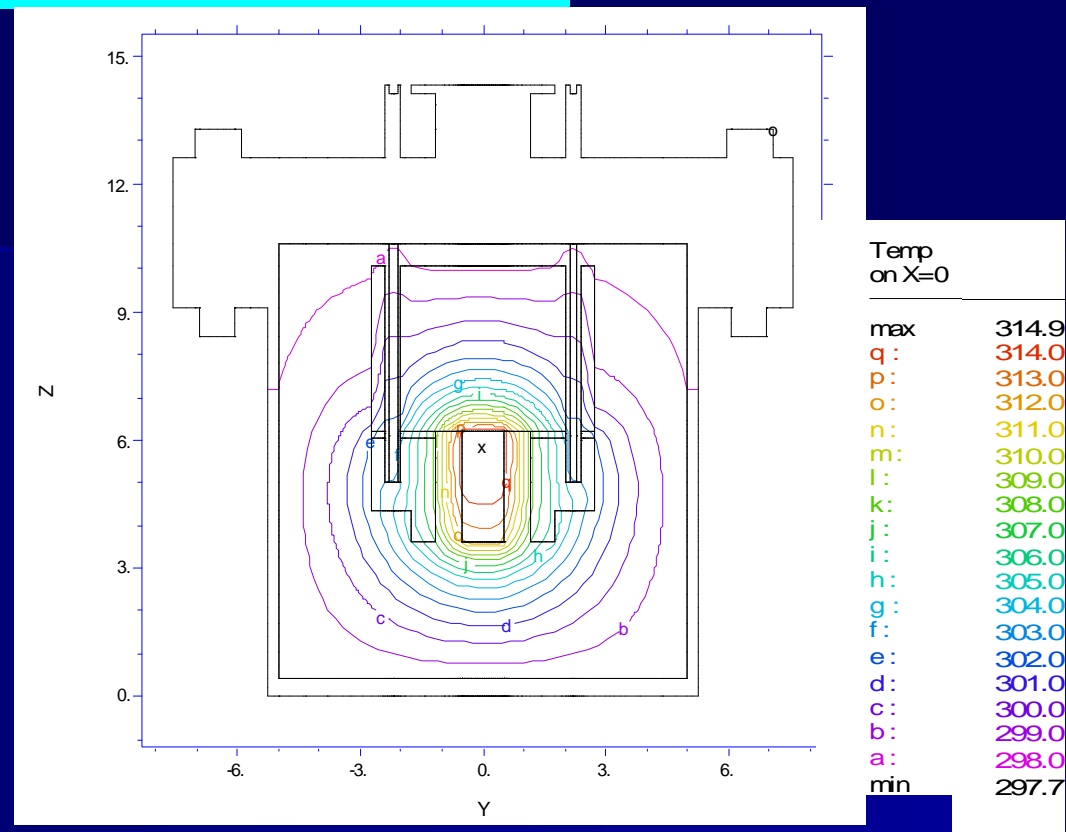
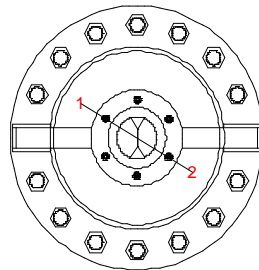
# Thermal Analysis



Temperature evolution

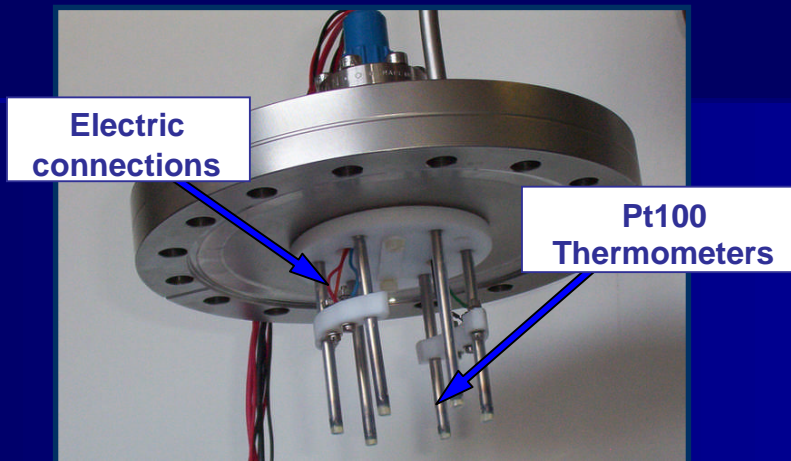


Temperature profile along 1-2

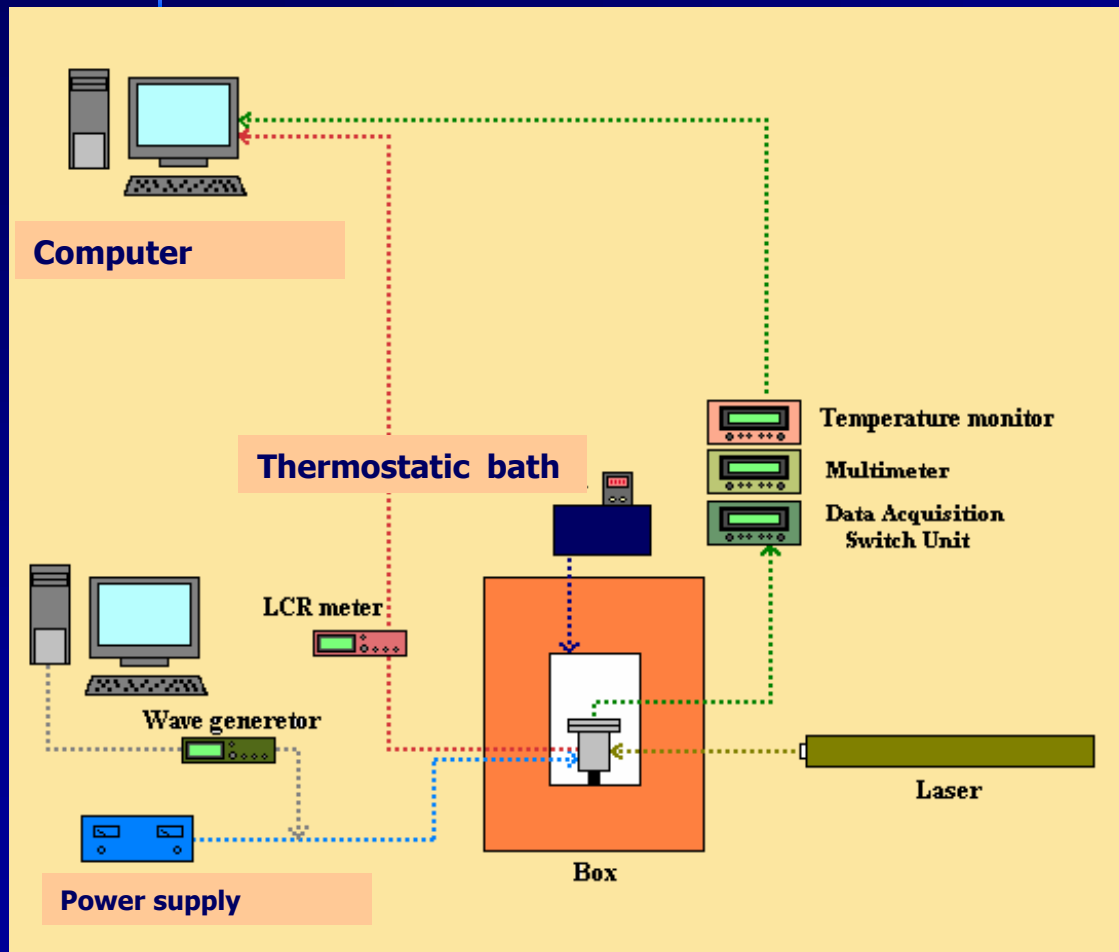


Isotherms axial plane

# Cell

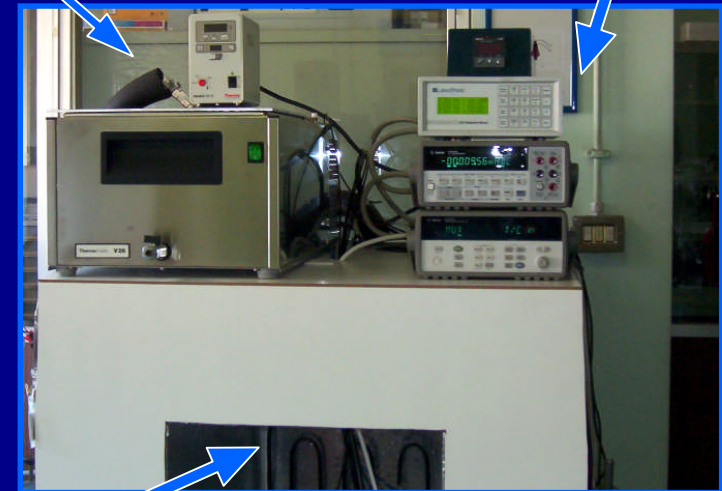


# Experimental System Set up



Thermostatic bath

Measurement instruments

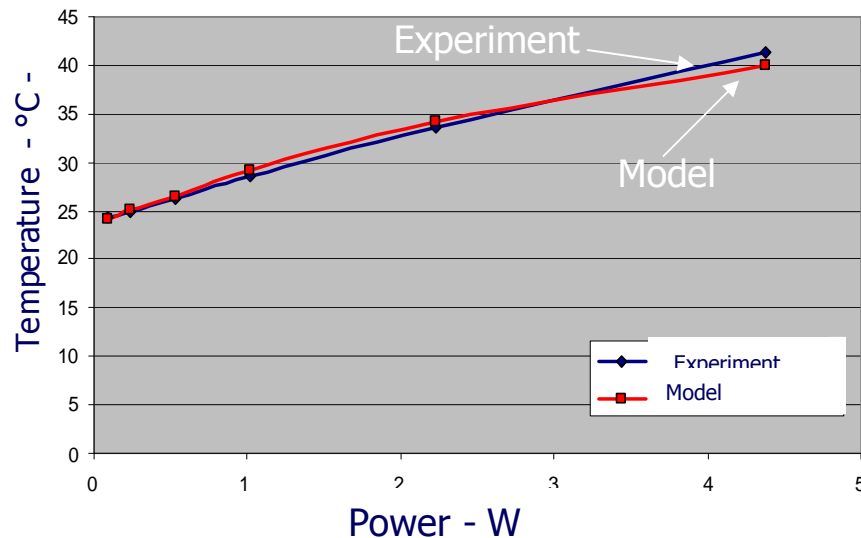


Thermostatic box

Thermostatic System

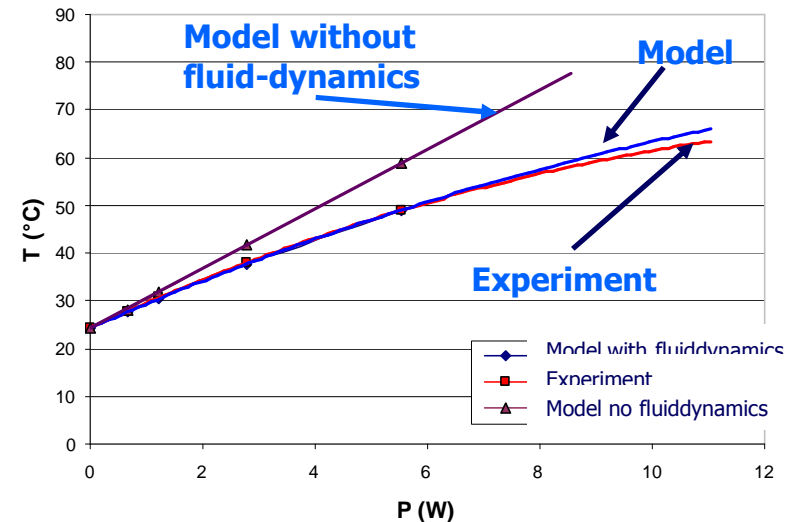
# Comparison between Experimental Data and Model

## Calibration and Model



Comparison between calibration data and model

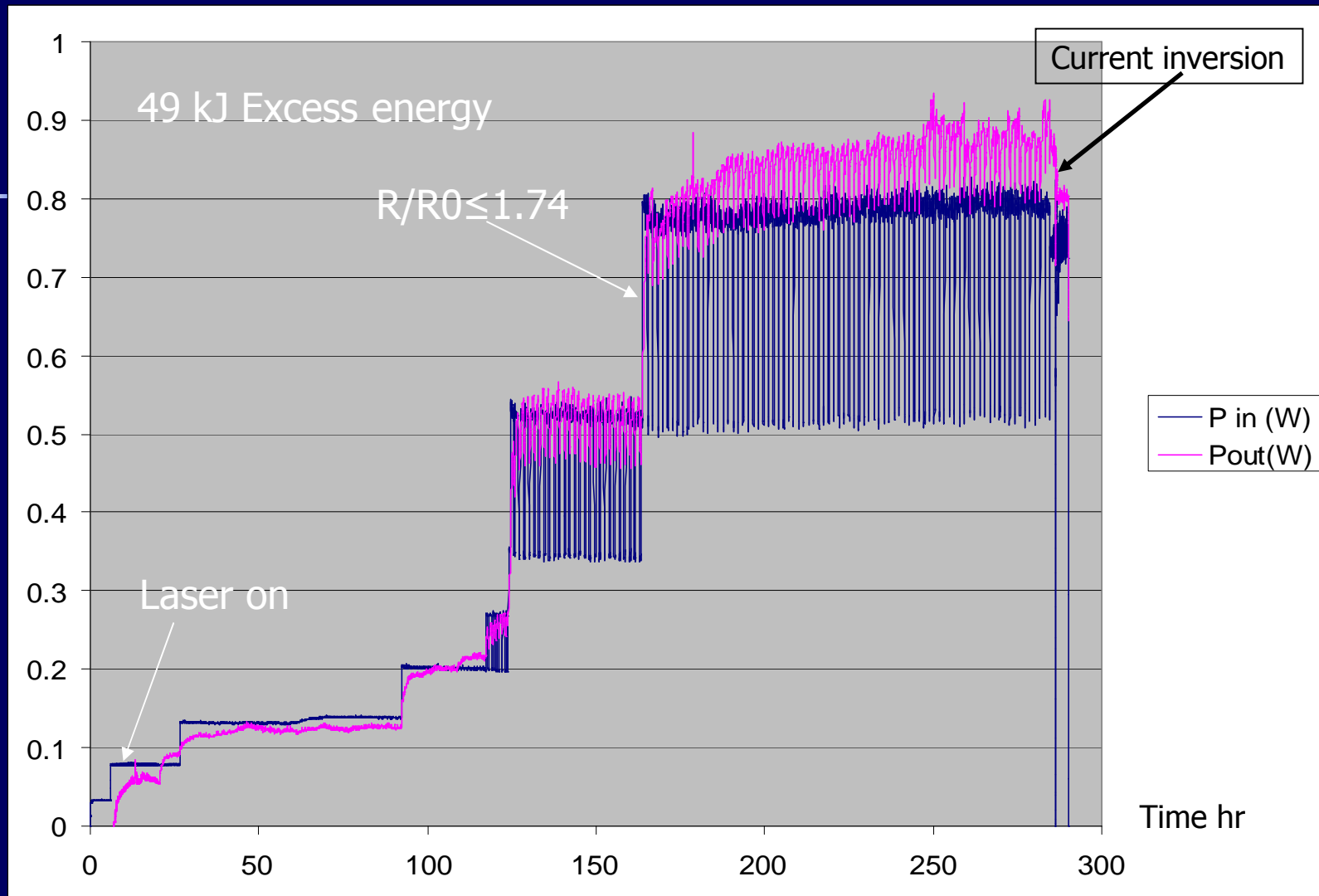
## Models and experiment



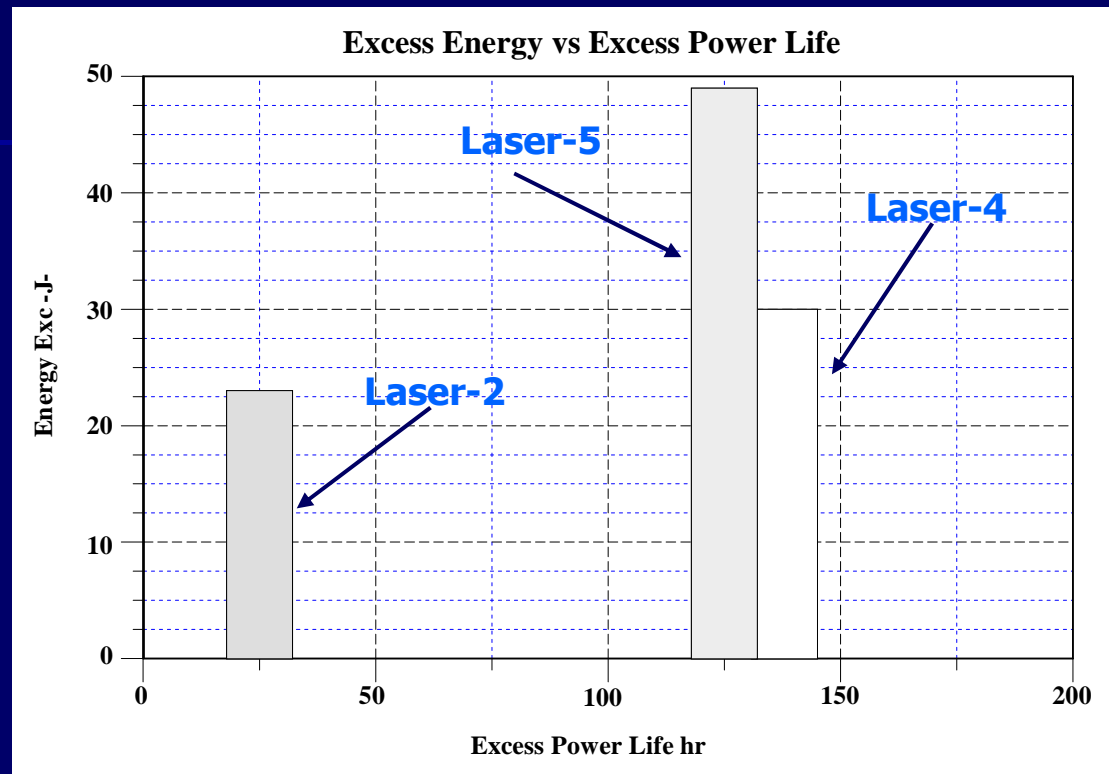
Comparison between experimental data and model, with and without fluid-dynamics



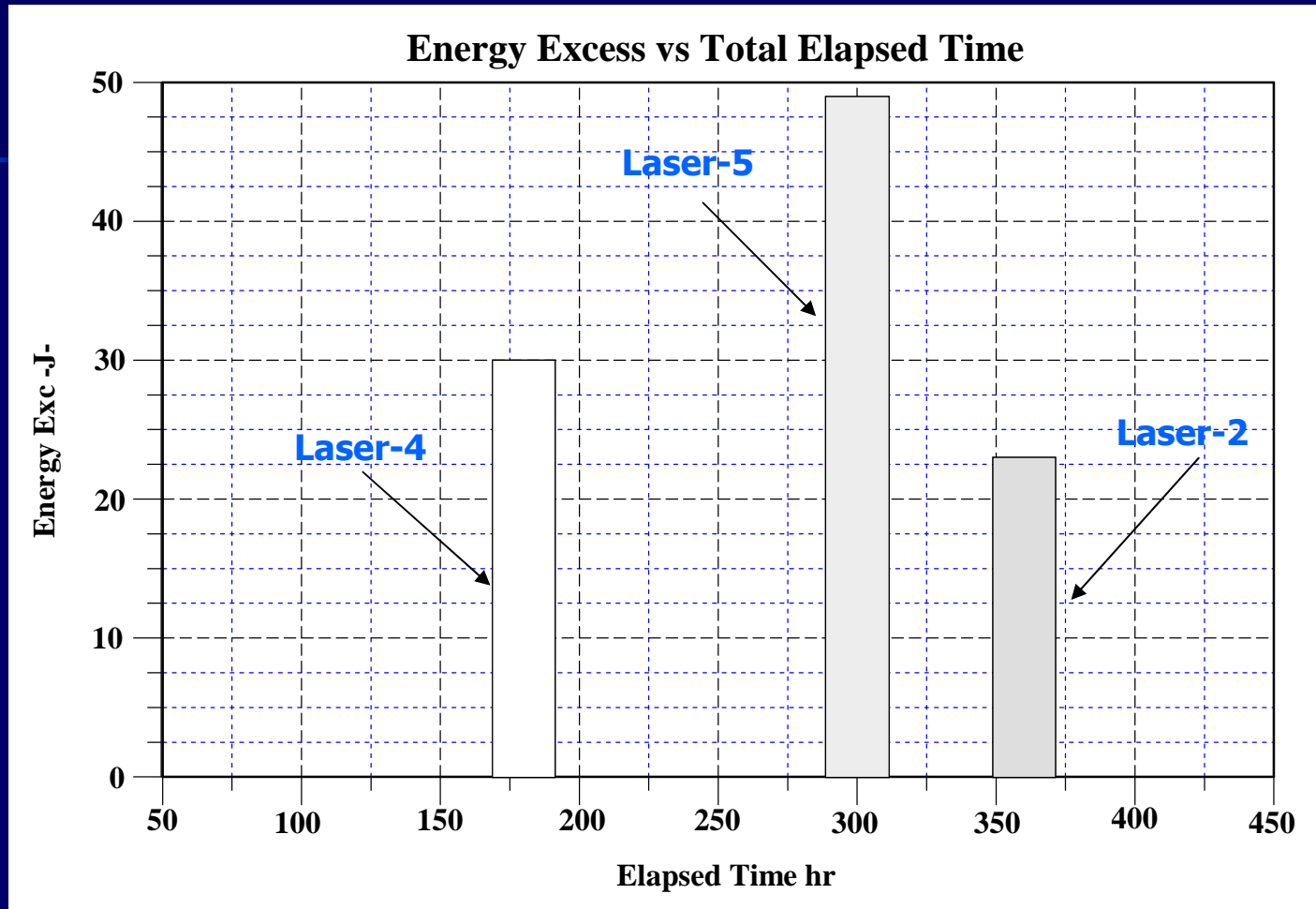
## Laser 5 Experiment: Calorimetric Results



Excess power during laser triggering (HeNe laser)



Excess energy vs excess power life time



Excess energy vs experiment elapsed time

# Conclusions

- **The improvement of the calorimeter design allowed to obtain a satisfactory agreement between model and experiment.**
- **Laser trigger gave significant reproducibility:  
excess power in 4 out of 5 experiments**
- **The amplitude of the effect is not yet under control  
even though high loading is almost always achieved.**
- **Material studies are in progress to identify and control  
the status of the system producing enhanced values of  
the signal.**